

Biphasic Hydroformylation of Higher Olefins

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AIChE Spring Meeting

New Orleans, LA

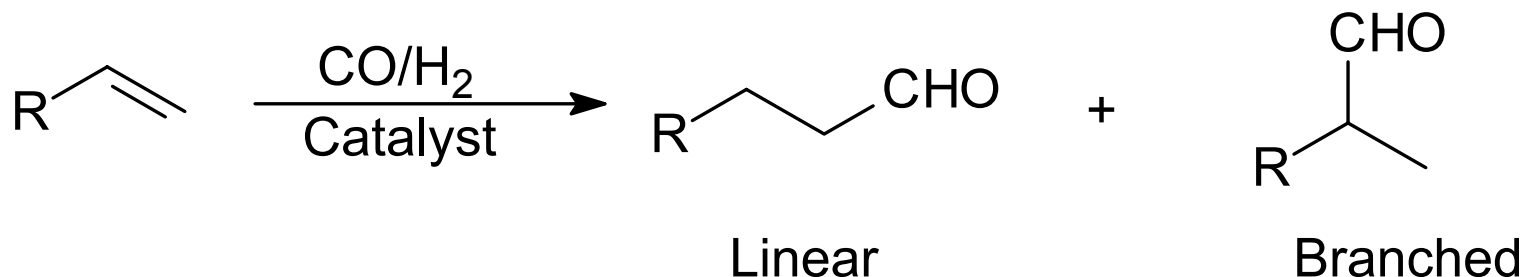
April 27, 2004

Project Summary

- **Objective: Develop improved hydroformylation process to produce higher aldehydes based on:**
 - Thermomomorphic solutions using conventional solvents
 - Rhodium phosphine catalysts
 - Higher olefins (octene through tetradecene)
- **Process produces aldehydes with good (high) ratio of linear to branched products**
- **Low loss of expensive rhodium catalyst during recycle**

Hydroformylation

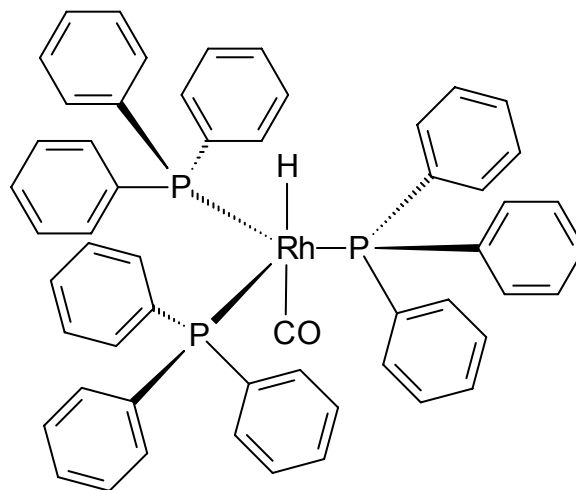
- **Largest commercial application of homogeneous catalysis**



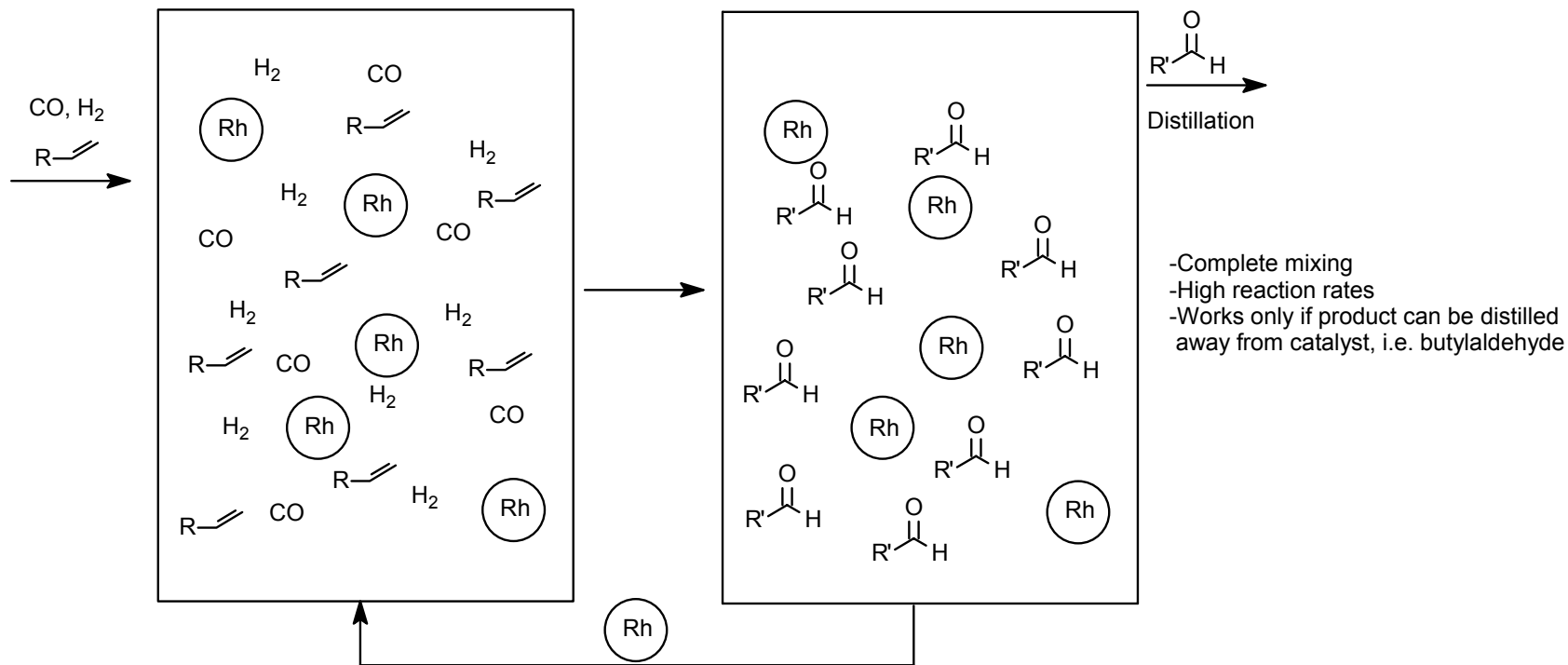
- **Worldwide production of aldehydes exceeds 7 million tons/year**
- **Higher aldehydes (>C₆) are important intermediates for solvents, detergents, surfactants, lubricants and plasticizers**

Aldehyde Production

- Rhodium and cobalt catalysts can be used
- Rhodium phosphine catalysts have replaced cobalt catalysts for lower aldehydes (<C₆) because of their higher activity and selectivity
- Because of the high cost of rhodium, catalyst losses have to be very low



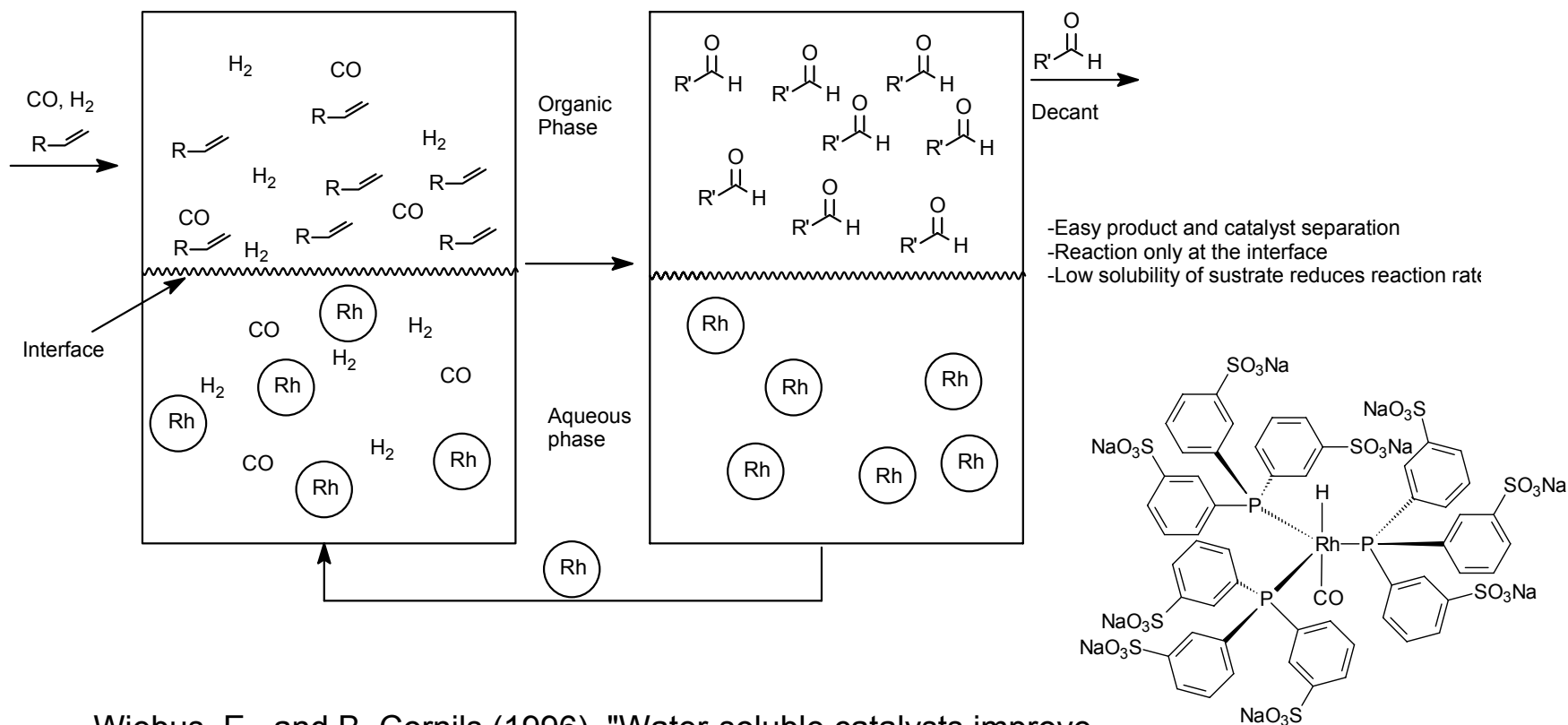
Single-Phase Homogeneous Hydroformylation (lower aldehydes)



Higher Aldehydes

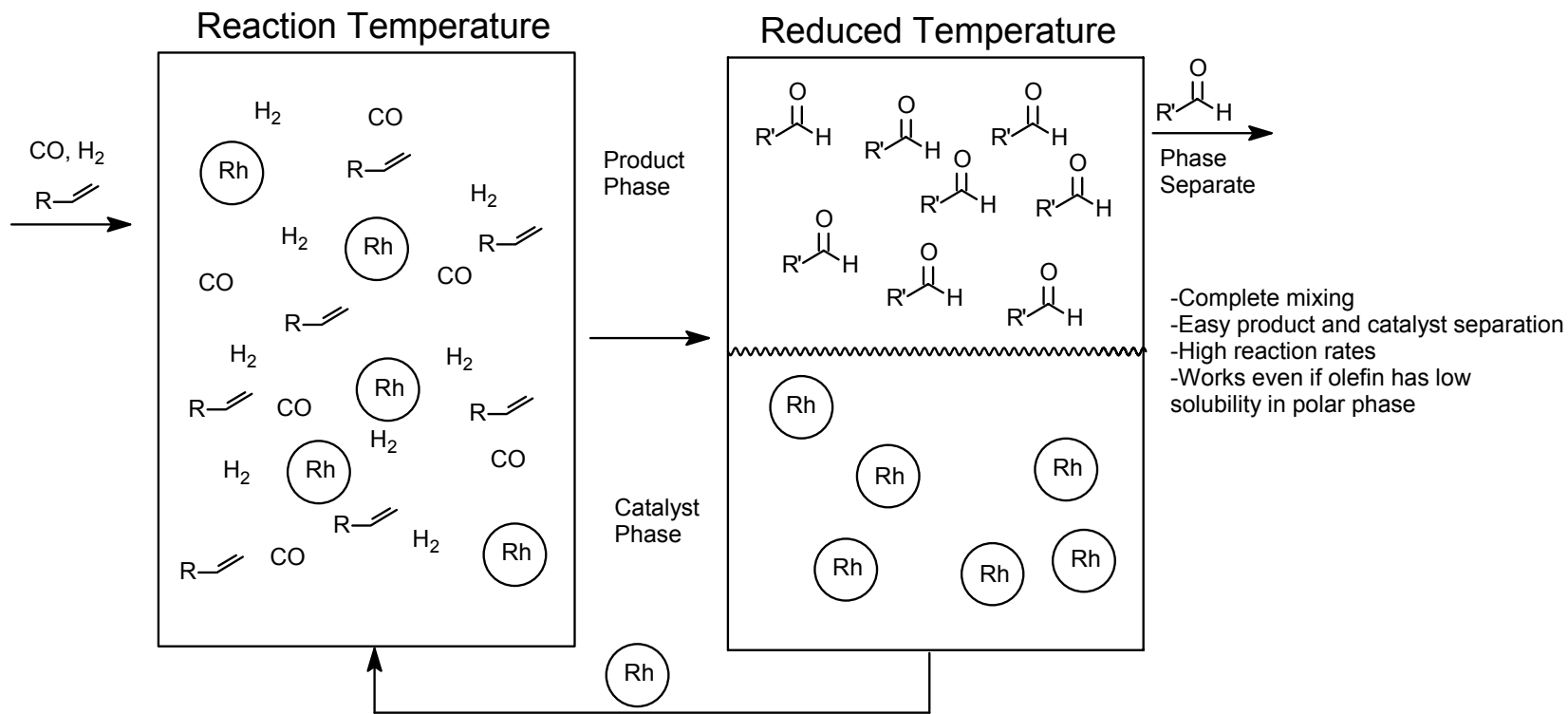
- **The higher boiling points of higher aldehydes result in catalyst decomposition**
- **Higher aldehydes are produced using less efficient cobalt catalyst because there is no effective means to separate the catalyst from the reaction mixture**

Biphasic Hydroformylation (Ruhrchemie-Rhône Poulenc)



Wiebus, E., and B. Cornils (1996). "Water-soluble catalysts improve hydroformylation of olefins," *Organic Processing*, March 63-66.

Thermomorphic Hydroformylation



Fluorous Thermomorphic Hydroformylation

- Hydroformylation has been done using perfluoromethylcyclohexane/toluene thermomorphic solution with fluorinated phosphine ligands
- Catalyst: $\text{HRh}(\text{CO})\text{-}(\text{P}(\text{CH}_2\text{CH}_2(\text{CF}_2)_5\text{CF}_3)_3)_3$
- Problem is that fluorinated solvents and ligands are very expensive

Horváth, I.T., and J. Rábai (1994). "Facile Catalyst Separation Without Water: Fluorous Biphasic Hydroformylation of Olefins," *Science* 266, 72-75

Non-Fluorous Thermomorphic Solutions

- **Common solvent pairs that reversibly change from being biphasic to monophasic as a function of temperature**
 - Methanol/cyclohexane (125°C)
 - *n*-butanol/water (49°C)
- **TDA's approach is to use conventional solvents and ligands for thermomorphic hydroformylation**

Thermomorphing Catalyst Solution



25°C



70°C

Thermomorphic Hydroformylation Results (Batch)

Sample	1-Alkene	Conversion (mol %)	L/B ratio
1	Octene	16	6.0
2	Dodecene	7	3.9
3	Tetradecene	3	4.2

Reaction conditions: Alkene/Rh (molar ratio) = 426, CO/H₂ (molar ratio) = 1, P(CO/H₂) = 500 psi, T = 100°C. Reaction time = 2h.

Reuse of Catalyst (Batch)

Cycle	Olefin Conversion (mol. %)	Linear (mol %)	Branched (mol %)	L/B Aldehyde Ratio	Rh Loss ($\mu\text{g/g}$)*
1	36.6	25.6	11.0	2.3	3
2	25.8	25.4	0.4	64	<1
3	27.9	27.9	0.0	-	<1
4	29.9	29.7	0.2	148	6
5	19.4	17.7	1.7	10	1

Reaction conditions: Alkene = 1-octene, CO/H₂ (molar ratio) = 1, P(CO/H₂) = 500 psi, T = 100°C, reaction time = 4h, *Rh loss (μg) per gram of organic phase solution.

Technical Barriers

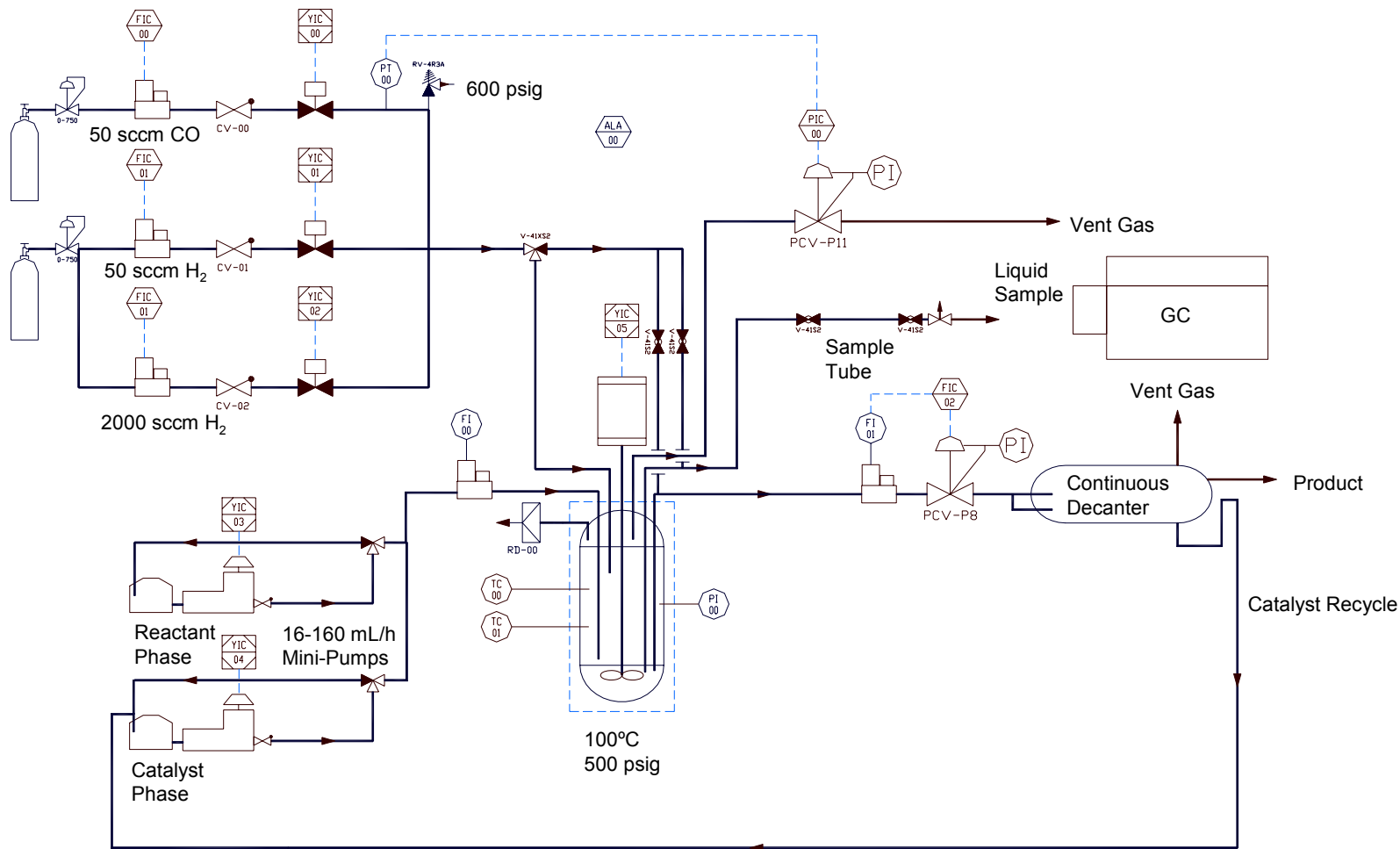
- **Show that our catalyst system can produce higher aldehydes in a continuous reactor with:**
 - High selectivity (linear/branched ratio)
 - Reasonable rates
 - Long catalyst lifetime
 - Low rhodium loss
- **How issue will be addressed**
 - Built a bench-scale continuous reactor
 - Do process optimization and long-term testing

Bench-Scale Continuous Reactor



- Reactor body manufactured by Autoclave Engineers
- 300 ml volume
- Fitted with Magnedrive stirrer and Dispersimax™ impeller for improved gas dispersion in liquid
- Rated up to 4700 psig at 200°C, 3800 psig at 538°C
- Computer controlled (OptoControl software)

P&ID of Bench-Scale Continuous Reactor



Summary

- **Successfully developed thermomorphic catalyst system for hydroformylation in batch mode**
- **Built a continuous bench-scale reactor**
- **In the process of optimizing catalyst system in a continuous system to be followed by long-term catalyst durability testing**
- **From the results of our work we expect to convince our commercial partner to license the technology**

Acknowledgement

- **This work is being funded by DOE SBIR Phase II project**
- **Contract: DE-FG03-01ER83307/A002**